This article was downloaded by: [Tomsk State University of Control Systems and Radio]

On: 19 February 2013, At: 11:30

Publisher: Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered

office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Molecular Crystals and Liquid Crystals Incorporating Nonlinear Optics

Publication details, including instructions for authors and subscription information:

http://www.tandfonline.com/loi/gmcl17

Deposition of Bi, Sr, Ca, and Cu Oxide Films by RF Glow Discharge Generated at Relatively High Pressure

Takuya Hashimoto ^b , Kunio Fukuda ^c , Masuhiro Kogoma ^d , Satiko Okazaki ^d , Mamoru Yoshimoto ^a & Hideomi Koinuma ^a

Version of record first published: 22 Sep 2006.

To cite this article: Takuya Hashimoto, Kunio Fukuda, Masuhiro Kogoma, Satiko Okazaki, Mamoru Yoshimoto & Hideomi Koinuma (1990): Deposition of Bi, Sr, Ca, and Cu Oxide Films by RF Glow Discharge Generated at Relatively High Pressure, Molecular Crystals and Liquid Crystals Incorporating Nonlinear Optics, 184:1, 201-205

To link to this article: http://dx.doi.org/10.1080/00268949008031762

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.tandfonline.com/page/terms-and-conditions

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae, and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand, or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

^a Research Laboratory of Engineering Materials, Tokyo Institute of Technology, 4259 Nagatsuta, Midori-ku, Yokohama, Japan, 227

^b University of Tokyo

^c Shin-Etsu Chemistry Co., Ltd.

^d Sophia University

Mol. Cryst. Liq. Cryst., 1990, vol. 184, pp. 201-205 Reprints available directly from the publisher Photocopying permitted by license only © 1990 Gordon and Breach Science Publishers S.A. Printed in the United States of America

DEPOSITION OF Bi, Sr, Ca, AND Cu OXIDE FILMS BY RF GLOW DISCHARGE GENERATED AT RELATIVELY HIGH PRESSURE

Takuya HASHIMOTO*, Kunio FUKUDA**, Masuhiro KOGOMA***, Satiko OKAZAKI***, Mamoru YOSHIMOTO, and Hideomi KOINUMA Research Laboratory of Engineering Materials, Tokyo Institute of Technology, 4259 Nagatsuta, Midori-ku, Yokohama, Japan 227 On leave from University of Tokyo, On leave from Shin-Etsu Chemistry Co., Ltd., On leave from Sophia University

<u>Abstract</u> Low temperature deposition of oxide films related to Bi-Sr-Ca-Cu-O superconductors has been investigated by using an apparatus originally designed for plasma chemical vapor deposition at moderately high pressures. Crystalline Bi_2O_3 film was obtained at a substrate temperature of $400\,^{\circ}\text{C}$ from triphenylbismuth. Amorphous films as-deposited at $400\,^{\circ}\text{C}$ from $Ca(DPM)_2$, $Sr(DPM)_2$, and $Cu(DPM)_2$ were annealed at $400\,^{\circ}\text{C}$ in air to be crystalline films of $CaCO_3$, $SrCO_3$, and CuO_3 , respectively. The effect of oxygen partial pressure on the crystallization of as-deposited films was examined with respect to $CaCO_3$ film to verify the effectiveness of elevated oxygen partial pressure for improving the film crystallinity. Linear relationships were obtained between the deposition time and film thickness. Preliminary data for the preparation of superconducting film by annealing a layered oxide film are also presented.

INTRODUCTION

Preparation of high- T_c superconducting oxide films by CVD were already reported, 1 but most of the reports employed thermal CVD using substrate temperatures higher than 700°C. For decreasing the film deposition temperature, we tried to use rf and microwave plasmas which could facilitate the decomposition of source gases and activate the oxygen. Recently, Kobayashi and coworkers reported the construction of $\text{Bi}_2\text{Sr}_2\text{CaCu}_2\text{O}_x$ crystal structure by a plasma assisted chemical vapor deposition at 650°C.^3 In these experiments except for ours, mixtures of source materials were introduced into the reactors to produce homogeneous superconducting films and the layer-by-layer deposition of component oxides has scarcely been attempted. This paper describes a low temperature ($<400^{\circ}\text{C}$) synthesis of Bi_2O_3 , SrCO_3 , CaCO_3 , and CuO_3 films using our originally designed CVD apparatus. Bi-Sr-Ca-Cu-O superconducting system was reported to have a structure of several numbers of Ca/CuO_2 bi-layers sandwiched by two $\text{Bi}_2\text{O}_2/\text{SrO}$ stacked

layers. For the artificial construction of these structures, we need to establish such a process technology as to enable the formation of as-grown crystalline film from each one of the sources at a low temperature and at a constant deposition rate.

EXPERIMENTAL

The setup of the apparatus used in the present study was reported elsewhere. 5 The condenser type electrodes structure was modified so that rf glow discharge could be sustained at a pressure even as high as latm; the cathode was tungsten needles with a diameter of 0.2mm and the anode was covered with a quartz plate of 0.2mm thick. The substrate MgO(100) was heated at 400°C . A mixture of plasma excited 0_2 and Ar gases was introduced above the substrate to be mixed with the source materials carried on helium flow. Helium was indispensable for sustaining the glow discharge at pressures higher than 10Torr. Rf plasma in a high oxygen partial pressure was expected to facilitate the preparation of oxide films efficiently. Source materials employed were $Bi(C_6H_5)_3$, $Sr(DPM)_2$, $Ca(DPM)_2$, and $Cu(DPM)_2$, where DPM represents $(CH_3)_3CCOCHCO(CH_3)_3$, supplied from Tri-chemical Co., Ltd. They were heated separately in oil baths and carried into the reaction chamber by helium. The evaporation temperatures were determined by TG measurement under Ar 100Torr atmosphere using a Sinku-Riko TGD-7000RH model. To prevent the source materials from condensation, the gas lines were heated up to 250°C. The film thickness was determined by the stylus method using a Taylor-Hobson Talystep. The crystal structure was analyzed by X-ray diffraction (XRD: MAC Science MXP³).

RESULTS AND DISCUSSION

Film deposition and crystal structure

Table 1 lists representative film preparation conditions and results. The as-deposited film prepared at $400\,^{\circ}\text{C}$ from $\text{Bi}(\text{C}_6\text{H}_5)_3$ has a crystal structure of Bi_2O_3 . Amorphous films were obtained from $\text{Sr}(\text{DPM})_2$, $\text{Ca}(\text{DPM})_2$, and $\text{Cu}(\text{DPM})_2$ at a reaction pressure of 3Torr. XRD Peak of CuO(111) appeared by annealing the film prepared from $\text{Cu}(\text{DPM})_2$ at $400\,^{\circ}\text{C}$ in air. Clear XRD patterns of CaCO_3 was observed in the film prepared from $\text{Ca}(\text{DPM})_2$ after annealing in air at $500\,^{\circ}\text{C}$. Annealing at a temperature of $400\,^{\circ}\text{C}$ also gave CaCO_3 pattern. Annealing the film pre-

pared from $\mathrm{Sr(DPM)}_2$ at 400°C also produced crystalline SrCO_3 . Sharper XRD pattern of SrCO_3 was observed by elevating the annealing temperature up to 600°C. Rather than CaO and SrO films, CaCO_3 and SrCO_3 films were preferentially formed probably because the carbonates were more stable thermodynamically under the experimental conditions and carbon was present in the source. Deposition at a higher oxygen partial pressure was effective to prepare as-grown crystalline films. Figure 1 shows the XRD patterns of the as-grown films prepared from $\mathrm{Ca(DPM)}_2$ at pressures of 3Torr and 10Torr. The film prepared at 10Torr showed CaCO_3 crystal structure without post-annealing.

Deposition rate

Figure 2(a)-(d) shows the relationship between the film thickness and reaction time. Fairly good linear relationships were obtained for all of the plasma CVD reactions to give the deposition rates of 1.30, 1.67, 1.25, and 1.50A/s for $\rm Bi_2O_3$, $\rm SrCO_3$, $\rm CaCO_3$ and $\rm CuO$ films, respectively. Since the thickness of each oxide layers in $\rm Bi\text{--}Sr\text{--}Ca\text{--}Cu\text{--}O$ superconductor is 2^{1} 3A⁴, the thickness of each oxide film must be controlled at that order for layer-by-layer as-grown superconductor. The deposition rates we obtained (1.25 1 1.67A) appear to be a little too high for the purpose of such layer-by-layer deposition.

Preparation of a superconducting film

Four films were deposited successively under the conditions of Runs 1, 2, 3, and 4 in Table 1 from $\mathrm{Bi}(\mathrm{C_6H_5})_3$, $\mathrm{Sr}(\mathrm{DPM})_2$, $\mathrm{Ca}(\mathrm{DPM})_2$, and $\mathrm{Cu}(\mathrm{DPM})_2$. Each layer was designed to have a thickness of 1000A. The stacked film was annealed at 850°C in air for 30min. Low temperature resistivity was measured by the conventional four probe method. $\mathrm{T_c}^{\mathrm{onset}}$ was observed at 80K. Peaks of $\mathrm{Bi_2Sr_2CaCu_2O_x}$ appeared in XRD pattern of the film. Further study is in progress for better control of film deposition using a computer controlled gas supplying system.

Conclusion

Crystalline $\mathrm{Bi}_2\mathrm{O}_3$, SrCO_3 , CaCO_3 , and CuO films were prepared by rf plasma assisted CVD at a temperature of $400\,^{\circ}\mathrm{C}$. Increase of oxygen partial pressure was effective to prepare as-grown crystalline films. The accumulation of these four layers of about 1000A thick each at

 $400\,^{\circ}\text{C}$ and subsequent annealing at $850\,^{\circ}\text{C}$ gave a film indicating a superconductivity at 80K.

Acknowledgment

The authors acknowledge Mr. M. Nakabayashi for experimental assistance. This study was supported by a Grant-in-Aid for Scientific Research on Chemistry of New Superconductors from Ministry of Education, Science and Culture.

References

1. For example, H. Yamane, H. Kurosawa, H. Iwasaki, T. Hirai, N. Kobayashi, and Y. Muto <u>Jpn. J. Appl. Phys. 28</u> L827(1989).

2. M. Yoshimoto, T. Hashimoto, T. Kosaka, K. Fukuda, S. Okazaki, M. Kogoma, T. Asakawa, M. Kawasaki, and H. Koinuma, <u>Proc. of the 6th Symp. on Plasma Processing</u>, Kyoto Japan, 364(1989).

 K. Kobayashi, S. Ichikawa, and G. Okada, <u>Chem. Lett.</u> 1989 1415(1989).

4. For example, R. M. Hazen, L. W. Finger, R. J. Angel, C. T. Prewitt, N. L. Ross, C. C. Hadidiacos, P. J. Heaney, D. R. Veblen, Z. Z. Sheng, A. El. Ali, and A. M. Herman, Phys. Rev. Lett. 60 1657 (1988)

A. EL. Ali, and A. M. Herman, Phys. Rev. Lett. 60 1657 (1988)
5. H. Koinuma, K. Fukuda, M. Kogoma, S. Okazaki, T. Hashimoto, M. Kawasaki, and M. Yoshimoto, Proc. of the 9th Int'l Symp. on Plasma Chemistry Pugnochiuso Italy, 1521(1989).

Table I Representative preparation conditions and results

	Source	Evaporation	Carrier	Reaction	Depo.	Crystal	structure
		temperature			-	•	400°C anneal
		(°C)		(Torr)	(A/s)	_	
#1	Bi(C ₆ H ₅)	130	40	3	1.30	Bi ₂ O ₃	_
	Sr(DPM) ₂	240	6	3	1.67	amorphous	$SrCO_3$
#3	Ca(DPM) ₂	220	6	3	1.25	amorphous	CaCO3
#4	Cu(DPM) ₂	140	40	3	1.50	amorphous	CuO
	Ca(DPM) ₂	220	6	10	1.25	CaCO3	_

rf power: 100W Substrate temperature: 400°C Ar: 2sccm 02: 2sccm

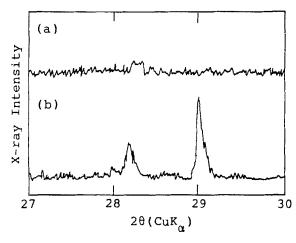


Fig. 1 X-ray diffraction patterns of the as-grown films prepared from Ca(DPM)₂ at a reaction pressure of (a) 3Torr and (b) 10Torr.

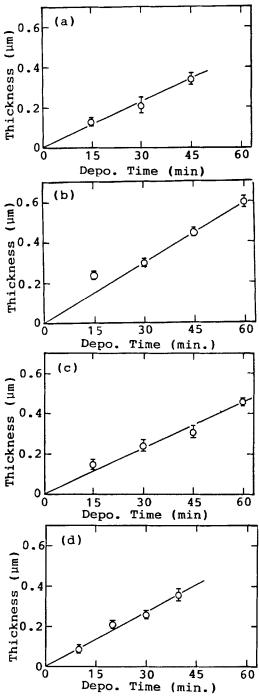


Fig. 2 Time dependence of film thickness prepared from (a) ${\rm Bi(C_6H_5)}$, (b) ${\rm Sr(DPM)_2}$, (c) ${\rm Ca(DPM)_2}$, and (d) ${\rm Cu(DPM)_2}$.